

Name: \_\_\_\_\_

Show work needed to justify your answer.

Date: \_\_\_\_\_

HW: # 20a: Math IBSL - Standard 20 - Applications of Arithmetic and Geometric Patterns

5 points

- 1 A high white blood cell count can indicate that the patient is fighting an infection. A doctor is monitoring the number of white blood cells in one of her patients after receiving antibiotics. The lab returns the following data.

Hour	0	12	24	36
White blood cells [cells mL <sup>-1</sup> ]	12 500	11 000	9680	8518.4

- a Create a general formula to model the patient's white blood cell count at any given time.
- b Use your general formula to calculate the number of white blood cells this patient will have after three days.  $\frac{72}{12} = 6$
- c Discuss the limitations of your general formula.

$$\frac{11000}{12500} = 0.88 \quad \frac{9680}{11000} = 0.88$$

$$(a) \quad U_n = 12500 (0.88)^t$$

t is time every 12 hours.

$$(b) \quad U_6 = 12500 (0.88)^6$$

$$U_6 = 5805$$

(c) white blood cell count will go back to normal after virus is gone.

- 3 Half-life is the time required for a substance to decay to half of its original amount.

- a A radioactive isotope has a half-life of 1.23 years. Explain what this means.
- b Write a general formula to calculate the amount remaining of the substance.
- c Use your GDC to sketch a graph of this situation.
- d If you start with a 52-gram sample of the isotope, how much will remain in 7.2 years?

(a) Every 1.23 years half of the original Amt is left

$$(b) \quad A = A_0 \left(\frac{1}{2}\right)^{\frac{t-1}{h}}$$

where A is Amount remaining after t years, A<sub>0</sub> is original Amount and h is half life.

$$(d) \quad A = 52 \left(\frac{1}{2}\right)^{\frac{7.2-1}{1.23}}$$

$$A = 1.58 \text{ grams}$$